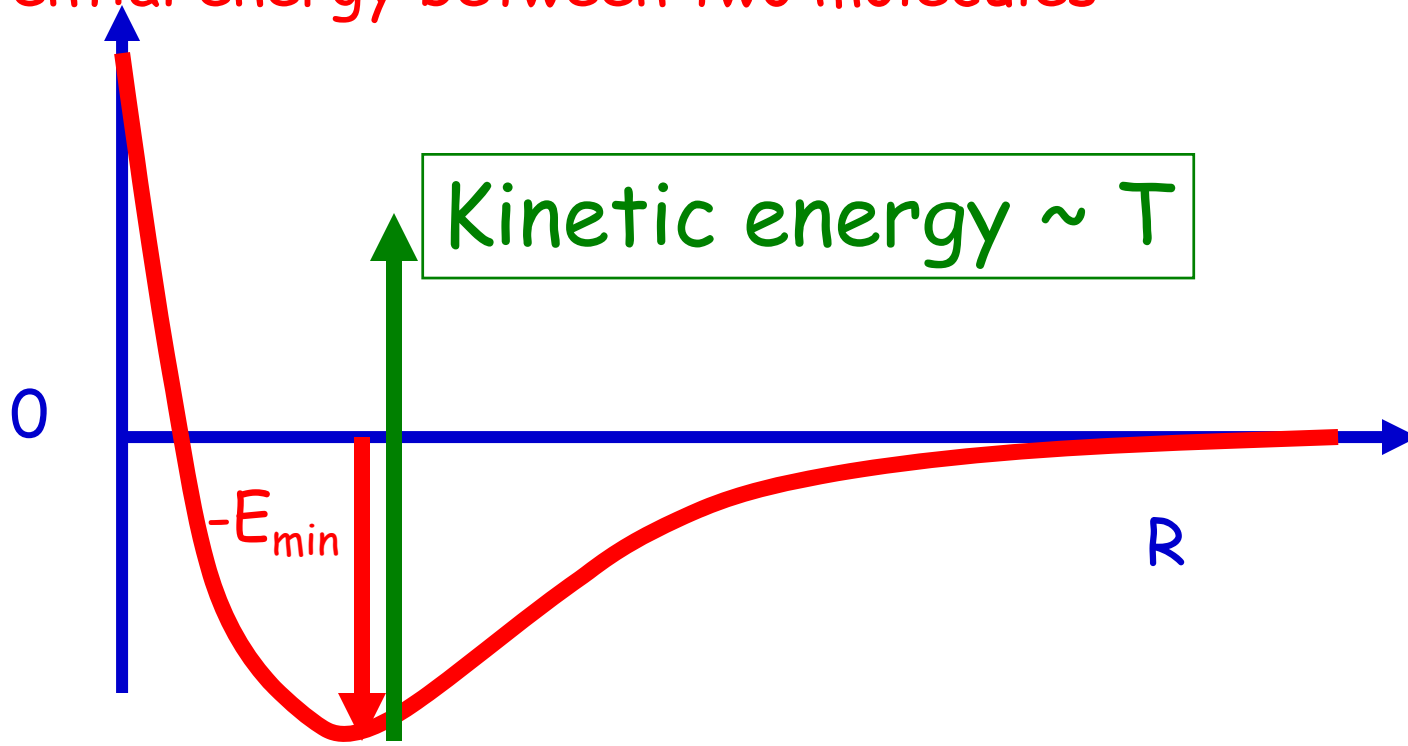
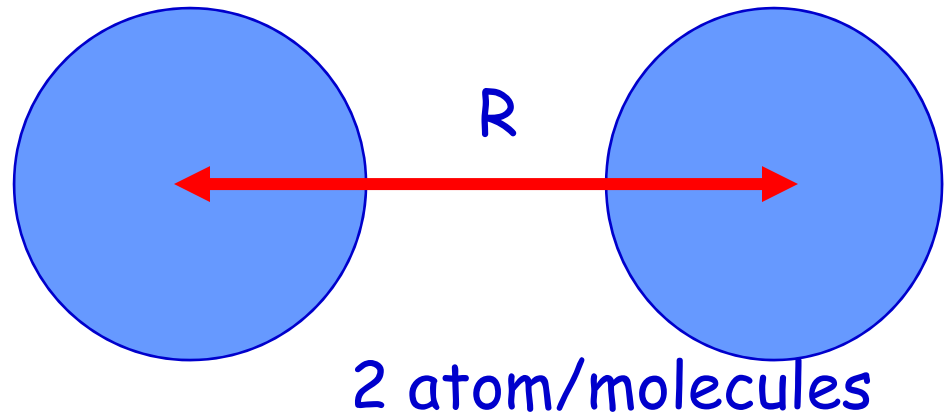


Potential energy between two molecules



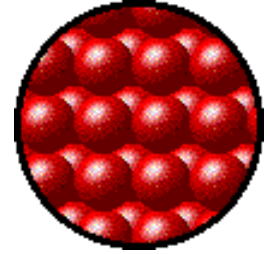
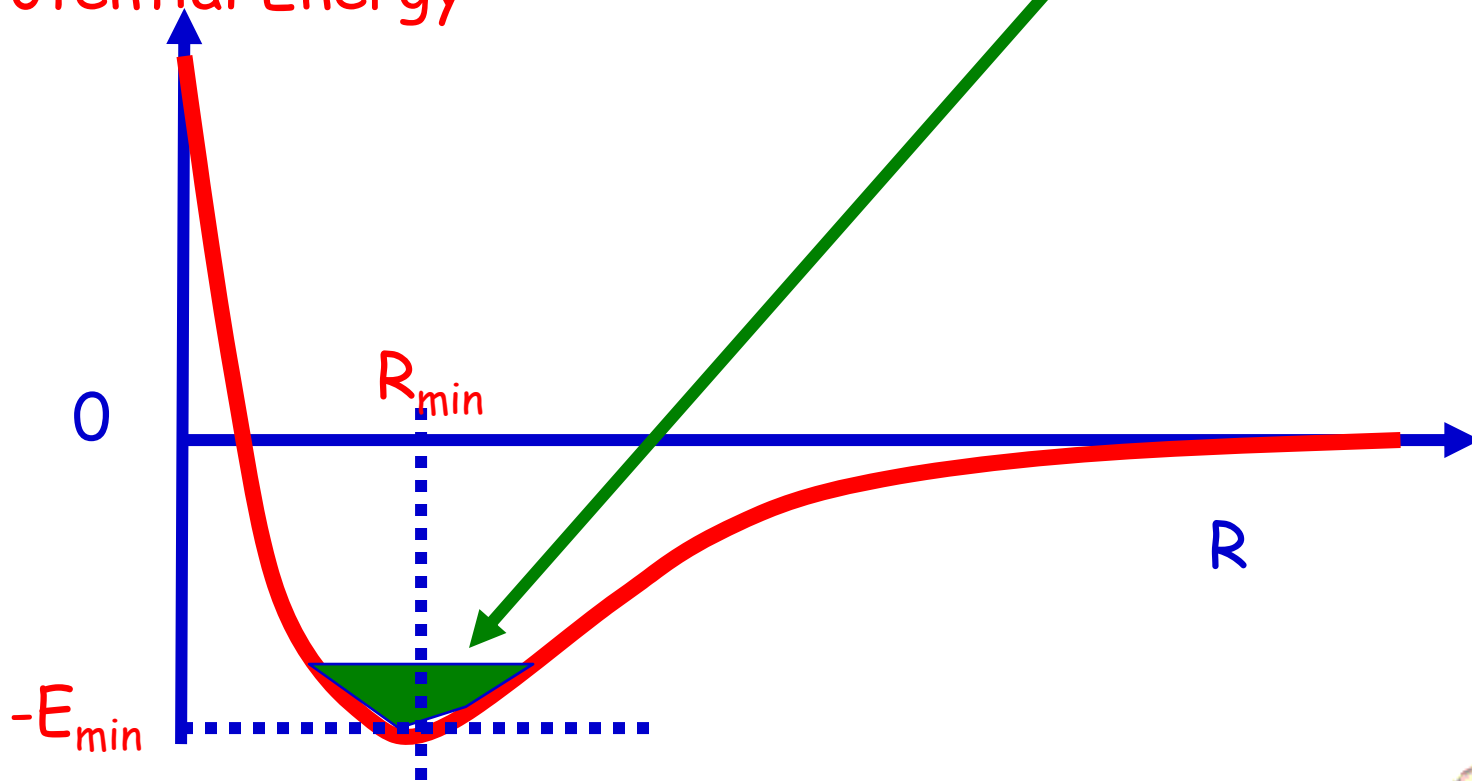
The curve depends on the material, e.g. E_{min} is different for water and iron



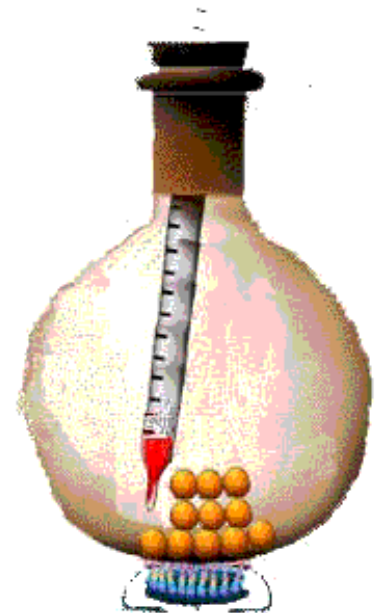
Solid (low T)

Kinetic energy $\sim T$

Potential Energy



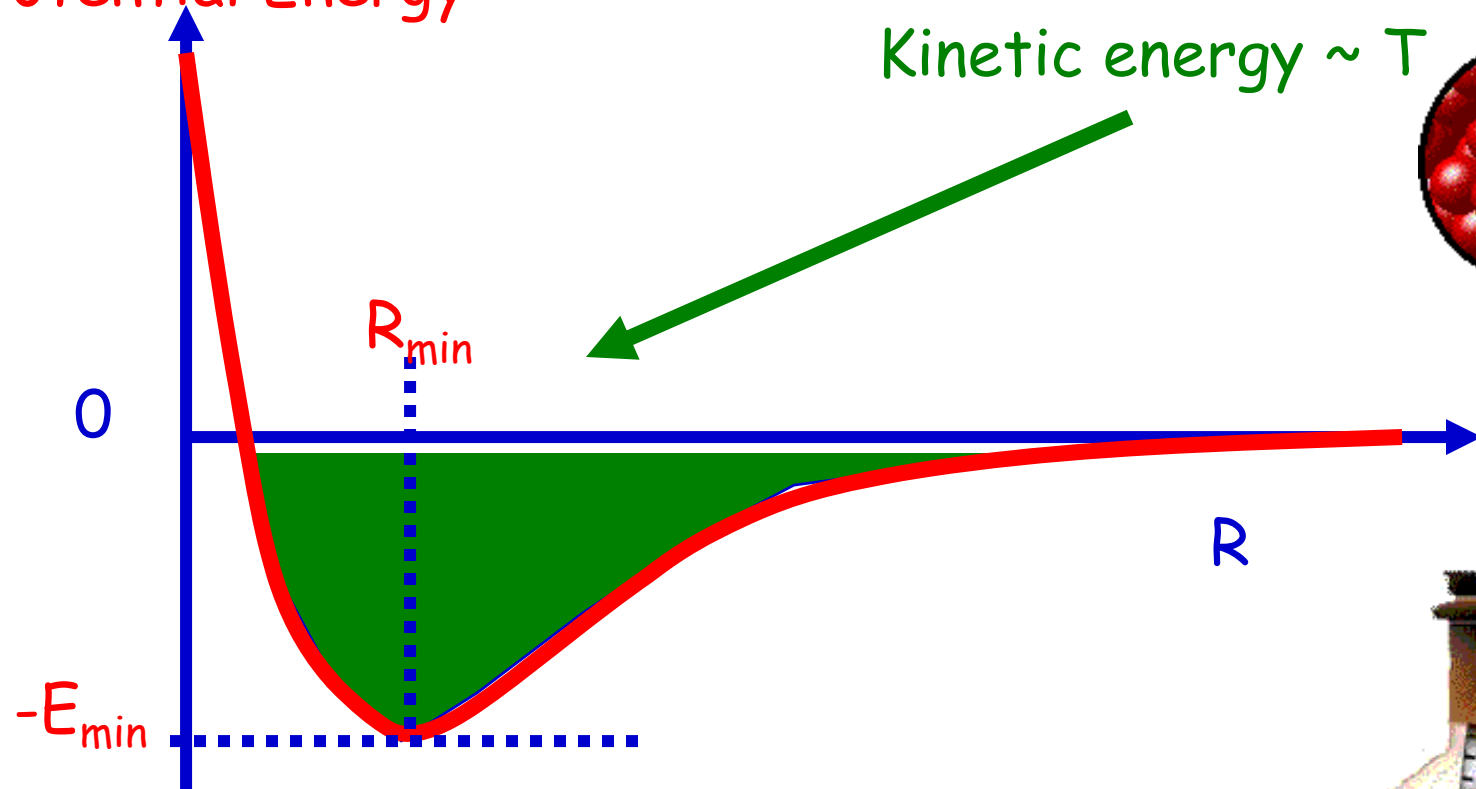
The temperature (and thus kinetic energy) is so small that the atoms/molecules can only oscillate around a fixed position R_{min}



Liquid (medium T)

Potential Energy

Kinetic energy $\sim T$



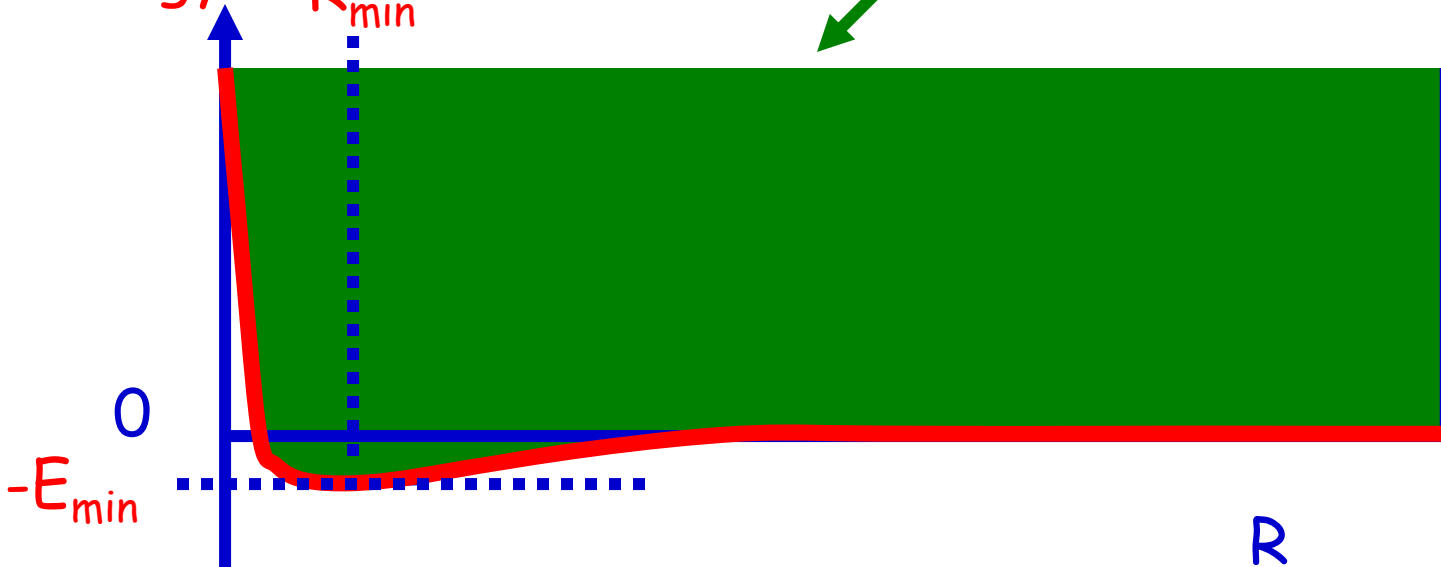
On average, the atoms/molecules like to stick together but sometimes escape and can travel far.



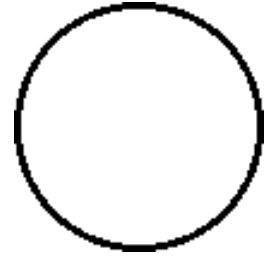
Gas (high T)

Kinetic energy $\sim T$

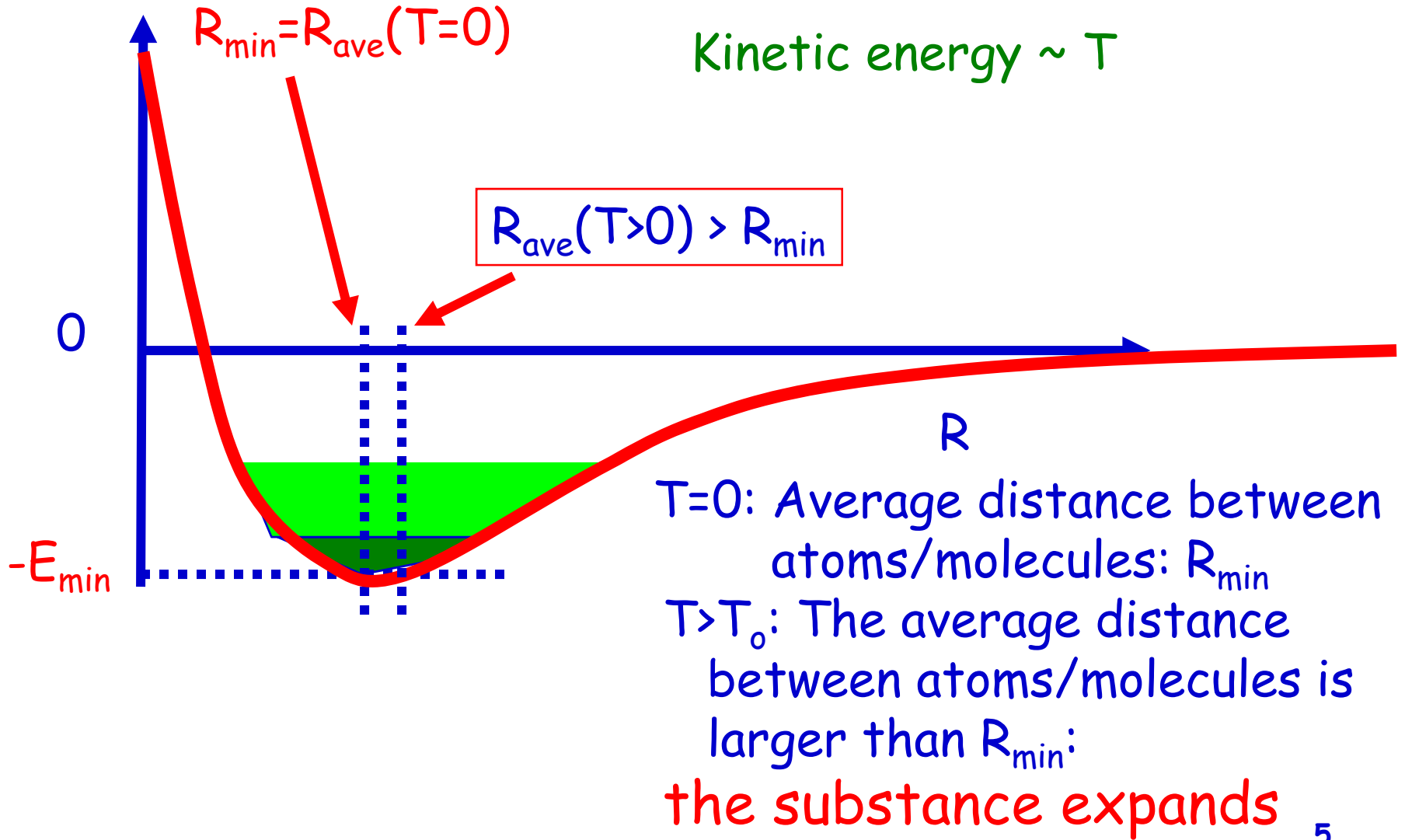
Potential Energy



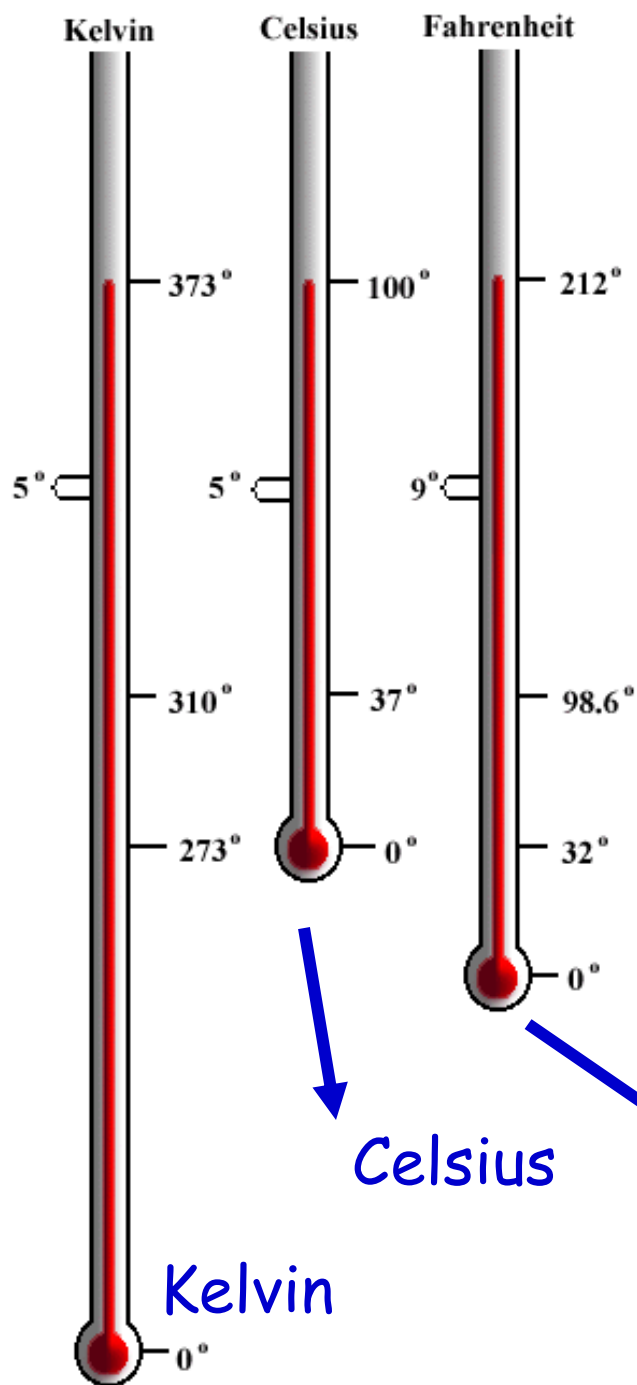
The kinetic energy is much larger than E_{\min} and the atoms/molecules move around randomly.



What happens if the temperature of a substance is increased?



Temperature scales



Conversions

$$T_{\text{celsius}} = T_{\text{kelvin}} - 273.$$

$$T_{\text{fahrenheit}} = 9/5 * T_{\text{celsius}} + 32$$

We will use T_{kelvin} .

If $T_{\text{kelvin}} = 0$, the atoms/molecules have no kinetic energy and every substance is a solid; it is called the absolute zero-point.

Thermal expansion

length

$$\Delta L = \alpha L_0 \Delta T$$

surface

$$\Delta A = \gamma A_0 \Delta T \quad \gamma = 2\alpha$$

volume

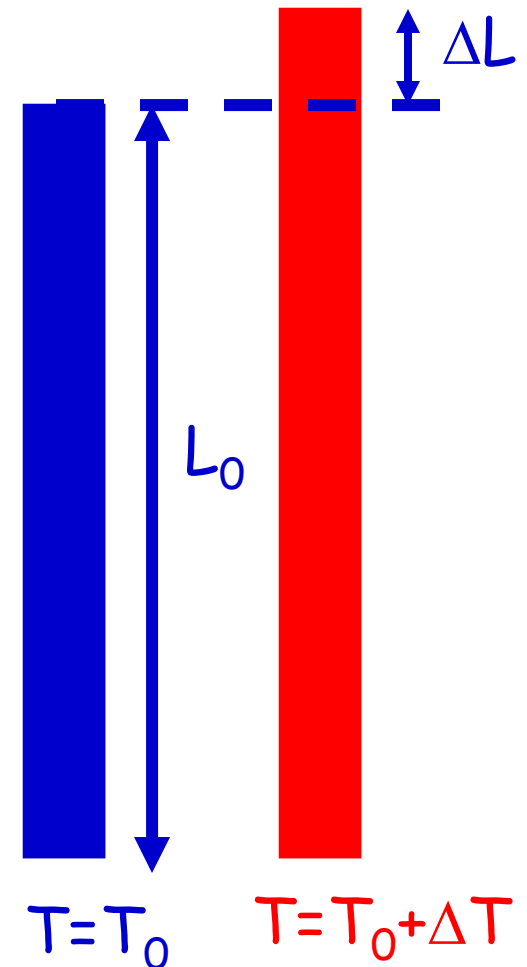
$$\Delta V = \beta V_0 \Delta T \quad \beta = 3\alpha$$

α : coefficient of linear expansion
different for each material

Some examples:

$\alpha = 24 \text{E-}06 \text{ /K}$ Aluminum

$\alpha = 1.2 \text{E-}04 \text{ /K}$ Alcohol

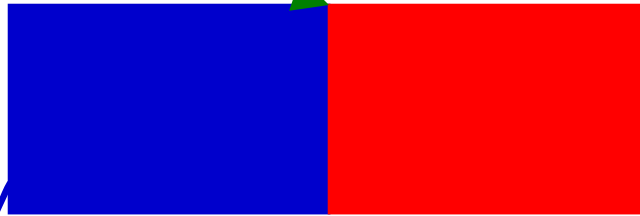


Thermal equilibrium

Thermal contact



Low temperature
Low kinetic energy
Particles move slowly



High temperature
High kinetic energy
Particles move fast

Transfer of kinetic energy



Thermal equilibrium: temperature is the same everywhere



Zeroth law of thermodynamics

If objects A and B are both in thermal equilibrium with an object C, then A and B are also in thermal equilibrium.



There is no transfer of energy between A, B and C

Thermal expansion: an example

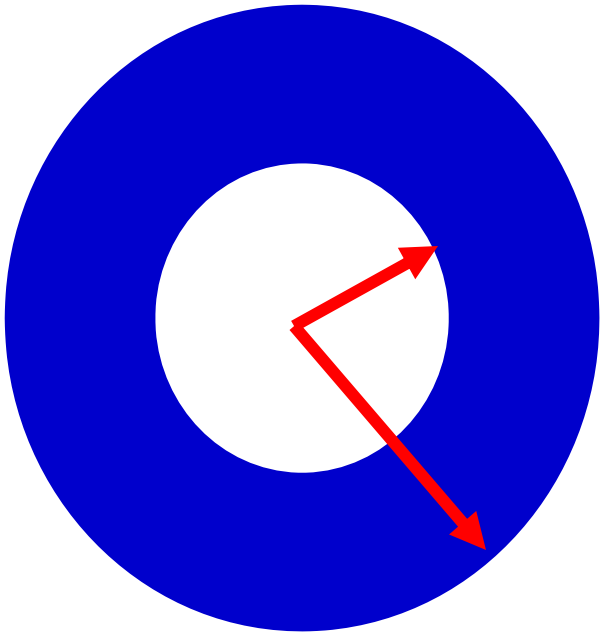
In the early morning ($T=30^\circ\text{F}=272.4\text{ K}$) a person is asked to measure the length of a football field with an aluminum measure and finds 109.600 m. Another person does the same in the afternoon ($T=60^\circ\text{F}=289.1\text{ K}$) using the same ruler and finds 109.566 m. What is the coefficient of linear expansion of the ruler?

$$\Delta L = \alpha L_0 \Delta T \quad \text{so} \quad \alpha = \Delta L / (L_0 \Delta T)$$

$$\Delta T = 16.7\text{ K} \quad L_0 = 109.60 \quad \Delta L = 109.600 - 109.566 = 0.034$$

$$\text{So: } \alpha = 24\text{E-06 /K}$$

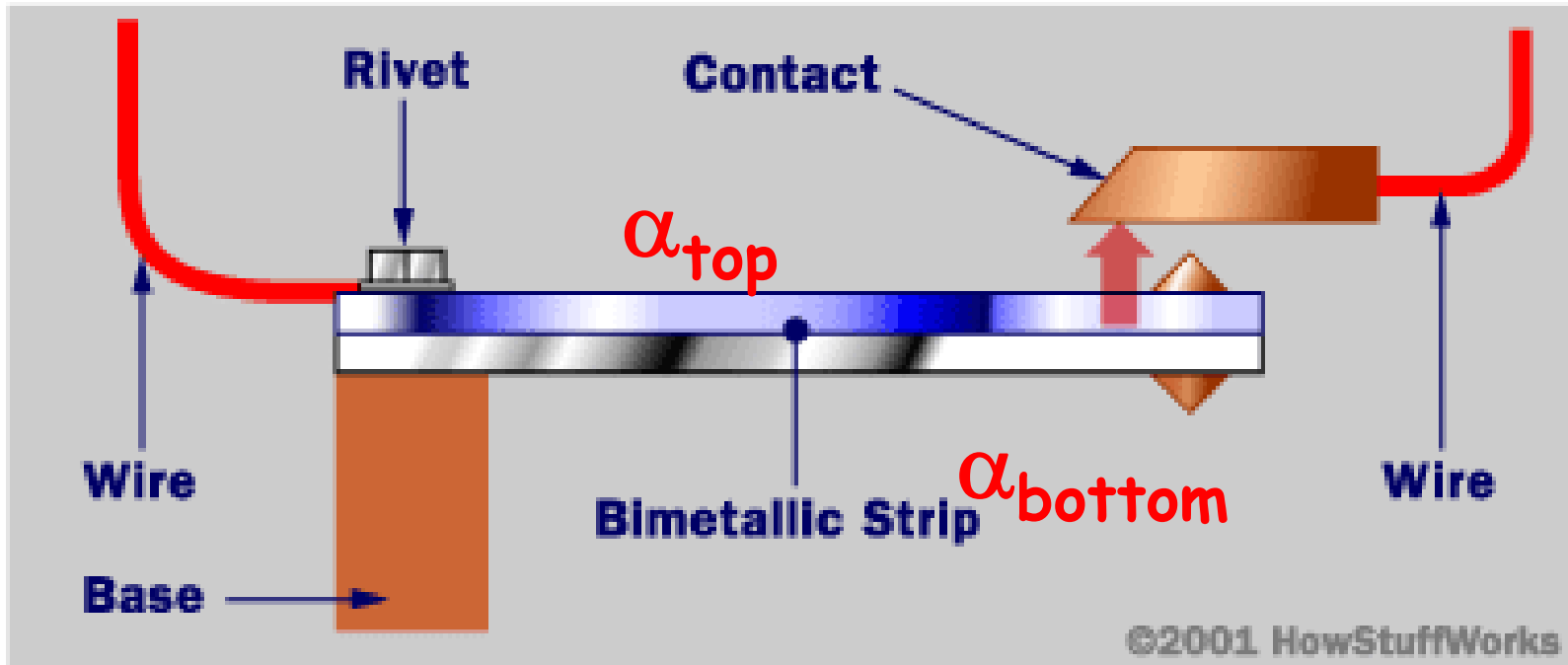
A heated ring



A metal ring is heated. What is true:

- a) The inside and outside radii become larger
- b) The inside radius becomes larger, the outside radius becomes smaller
- c) The inside radius becomes smaller, the outside radius becomes larger
- d) The inside and outside radii become smaller

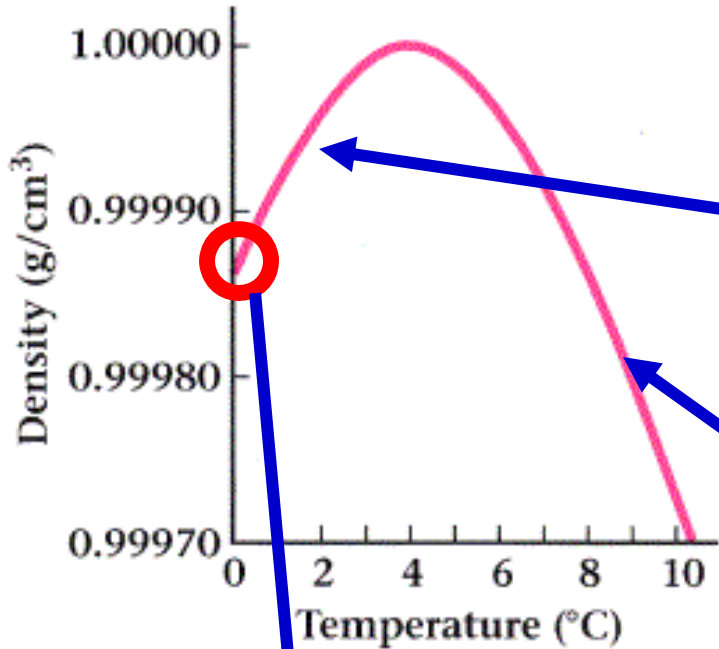
Bimetallic strips



Application: contact in a refrigerator

$\alpha_{top} < \alpha_{bottom}$ if the temperature increases,
The strip curls upward, makes contact and switches
on the cooling.

Water: a special case

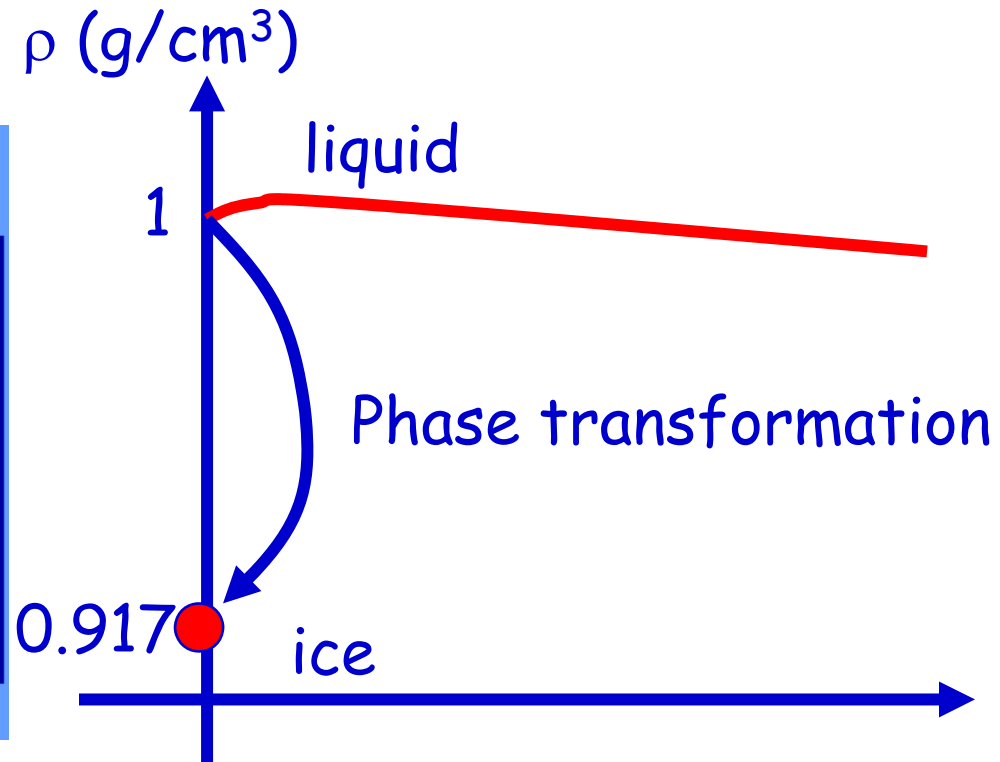
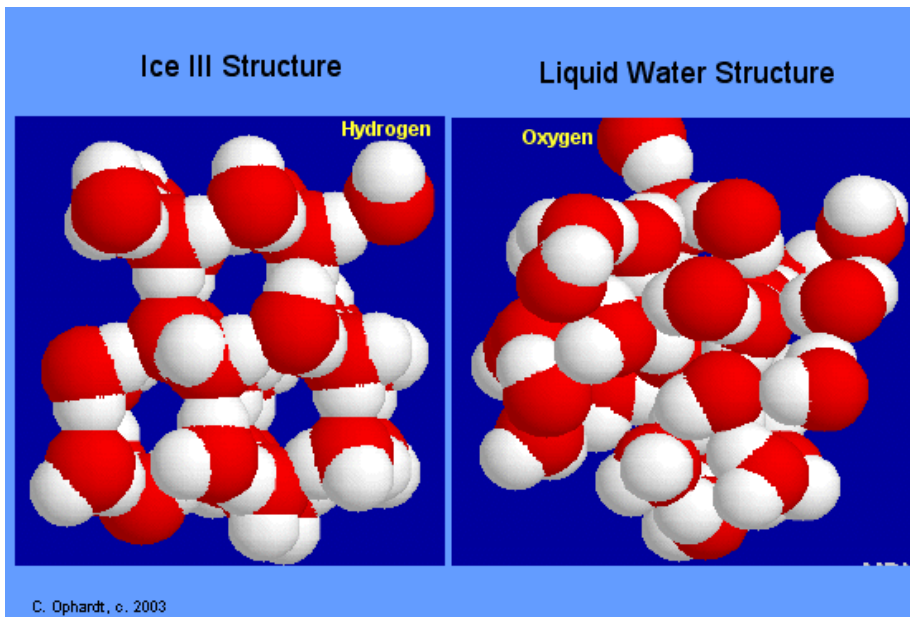


Coef. of expansion is negative: If T drops the volume becomes larger

Coef. of expansion is positive: if T drops the volume becomes smaller

Ice is formed (it floats on water)

Ice



Ice takes a larger volume than water!